# **Conformationally Constrained Nicotines.** 1-Pyridinyl-7-azabicyclo[2.2.1]heptane and 1-Pyridinyl-8-azabicyclo[3.2.1]octane Analogues

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Conformationally constrained bicyclo analogues of natural (-)-nicotine and unnatural (+)-nicotine have been synthesized from D- and L-glutamic acid, respectively. Regioselective addition of 3-lithiopyridine to the  $\gamma$ -carbonyl of a protected glutamate was followed by intramolecular imine formation and stereospecific catalytic hydrogenation of the resultant pyrroline to give *cis*-5pyridinylproline. A sequence of transformations to convert the ester to bromide was followed by the key intramolecular anionic cyclization at the benzylic position to form the 1-pyridinyl-7azabicyclo[2.2.1]heptane analogue. Alternatively, homologation of the ester of cis-5-pyridinylproline and conversion to bromide allowed cyclization to the 1-pyridinyl-7-azabicyclo[3.2.1]octane analogues.

#### Introduction

Nicotinic acetylcholine receptors are a group of ion channel receptors<sup>1</sup> that play an important role in many biological processes related to a number of nervous system disorders. As ligands for the nicotinic acetylcholine receptor,<sup>2,3</sup> many groups of compounds have been isolated from natural sources or synthesized, of which (–)-nicotine (**1**, Chart 1) and its analogues represent an important class. In particular, nicotine analogues with rigid structures have become interesting targets stimulated by the isolation of the very active but highly toxic epibatidine<sup>4</sup> from the skin of a South American frog.

The targets in the work we are now reporting have been to synthesize nicotine analogues 2 and 3, which possess azabicyclo ring systems with an apical nitrogen. The synthetic plan involves coupling of glutamic acid derivatives with 3-metalated pyridine, followed by cyclization to form proline A. By subsequent side-chain manipulation, A can afford bromide B or C (Figure 1). The key step then becomes generation of the carbanion at the benzylic position  $\alpha$  to the pyridine followed by intramolecular cyclization to construct the bridged system. Depending on the number of carbons in the side chain of the bromide, either a [2.2.1]- or [3.2.1]azabicyclonicotine analogue will result. Although Beak's pioneering work has presented many examples of metalation and electrophilic substitution at the  $\alpha$ -position of an amine derivative<sup>5,6</sup> (e.g., Boc-protected secondary amines), there were certain unexplored factors introduced by our system. Thus, there has been no report of application of this method to pyridines or to the deprotonation of a tertiary carbon  $\alpha$  to the nitrogen. Also, after deprotonation, although an intramolecular cyclization seems to be preferred judging from the ring size, steric effects at the quaternary carbon in forming the bicyclo system might

lead to competitive intermolecular reaction, including quaternization of the pyridine.

## **Results and Discussion**

Synthesis of the Pyridinyl Ketone. Using a number of glutamic acid derivatives and 3-metallopyridines, a variety of conditions and combinations were examined for pyridinyl ketone formation. The first substrate was *N-tert*-butyloxycarbonyl-L-glutamic acid α-*tert*-butyl ester  $(\mathbf{4})^7$  as its tetrahydrofuranyl ester.<sup>8</sup> In its reaction with the original Grignard reagent derived from 3-bromopyridine, a disappointing 30% yield of pyridinyl ketone 8 was obtained.

Since a quite satisfactory yield of the corresponding anisolyl ketone had resulted when the isoxazolidide 5 was treated with anisolyllithium,7 this combination was applied to the preparation of the pyridinyl ketone. With 3-pyridinyllithium<sup>9,10</sup> and isoxazolidide 5, an improved yield of 69% of ketone 8 was realized. Substitution of the 3-pyridinyl Grignard reagent led to a decreased yield of 50%.

The most convenient and efficacious process was found to be the coupling of 3-pyridinyllithium with the readily available *N*-Boc pyroglutamate esters<sup>11</sup> **6** and **7**. With *tert*-butyl ester **6**, a 73% yield of pyridinyl ketone **8** was realized, while methyl ester 7 gave the corresponding

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Chart 1 1. (S)-Nicotine 2 (-)-3 (+)-3

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Synthesis of Pyridinyl Ketones

Figure 1. Proposed synthesis of the [2.2.1] and [3.2.1] azabicyclonicotine analogues.

Scheme 1.



7: R = CH<sub>3</sub>

ketone **9** in 59% yield. These results are collated in Scheme 1.

Intramolecular Cyclization to the Proline Derivatives. Our initial intention was to selectively remove the *N*-Boc group without affecting the *tert*-butyl ester of **8**.<sup>12</sup> The free amine and the ketone functionalities thus could undergo intramolecular condensation to form the pyrroline under treatment with mild base, and subsequent reduction would produce the proline derivative. Removal of the N-Boc group, however, was not sufficiently selective and a partial removal of the tert-butyl ester was observed under various conditions. As a result, the plan was to remove both the N-Boc and tert-butyl ester groups under more strongly acidic conditions, then to re-esterify the acid. This was accomplished by treatment of 8 with saturated HCl in MeOH from which a 75% yield of keto-methyl ester 10 was obtained. A major side product was the ketal-methyl ester 11, which could be easily hydrolyzed to ketone 10 at pH 1-2 with aqueous HCl. On the other hand, when methyl ester 9 was used, treatment with 3 M HCl in EtOAc afforded the single product 10. After isolation using base, the pyrroline 12 was obtained in 87% yield. Catalytic hydrogenation of the imine gave the *cis*-proline derivative 13 as the sole product in 95% yield. In another experiment, with NaCNBH<sub>3</sub> as the reducing agent, a mixture of both diastereoisomers with a cis to trans ratio of 5/4 was isolated, as determined by <sup>1</sup>H NMR. These experiments are summarized in Scheme 2.

**Side Chain Extension.** With pyridinyl proline **13** in hand, the next task was to extend the carboxylate side chain at the 5-position. For this purpose, the nitrogen was protected as its Boc derivative and the methyl ester was hydrolyzed with LiOH to the acid **16**. A stepwise route utilizing the Arndt–Eistert synthesis<sup>13,14</sup> was applied to obtain the homologated ester **18**. In this sequence, a mixed anhydride intermediate<sup>15</sup> was superior to the acid chloride for diazoketone formation. Reduction of the ester with  $Ca(BH_4)_2$  afforded both the desired alcohol **19** plus the borane complex of **19**. Use of a large amount of EtOH as solvent instead of a mixture of THF/ EtOH<sup>16</sup> suppressed formation of this complex, which can also be broken up by refluxing in EtOH with 400 mol % of diethanolamine.

Our intention was to obtain bromide **21** directly from alcohol **19** using carbon tetrabromide and triphenylphosphine;<sup>17</sup> however, only poor yields of **21** were isolated. An alternative was to prepare the mesylate **20** and to use it directly in the alkylation step or convert it to bromide **21** as the alkylation reagent. All attempts at standard mesylate formation, requiring base, failed; elimination was the major product. A milder way to effect mesylation under essentially neutral conditions is to use the very active mesylating reagent 2-mesyl-5-methylimidazonium triflate.<sup>18</sup> When this reagent was used mesylate **20** was the sole product and treatment with LiBr in acetone afforded bromide **21** in 73% yield. A minor and

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Scheme 2. Synthesis of 5-Substituted Proline Derivatives



unexpected product was cyclic carbamate **22**, resulting from attack at the carbonyl oxygen of the Boc group by the very active mesylated side chain during bromide formation. Bromide **21** was not stable, due to alkylation of the pyridine nitrogen by the side chain bromide. This is probably an intermolecular process since dilution with solvent significantly slows this process. These reactions are depicted in Scheme 3.

**Construction of the Bicyclo [2.2.1] Ring System.** Anion formation and closure of bromide **21** to the bicyclo [2.2.1] ring system was surprisingly sensitive to the specific base, as shown in Scheme 4. When KHMDS was used, a small amount of olefin **23** was isolated together with recovered starting material **21**. The structure of the olefin was confirmed by an alternative synthesis: methyl ester **15** was reduced with Ca(BH<sub>4</sub>)<sub>2</sub>, and the resulting alcohol **25** was oxidized to the aldehyde **26** and treated with methyltriphenylphosphonium bromide/*n*-Buli to afford the olefin **23**. When *s*-BuLi was used, two products were obtained, the desired [2.2.1] structure **2a** and **24**, in which the bromine had been replaced by hydrogen. The structure of **24** was established by catalytic hydrogenation (Pd/C) of olefin **23** to **24**. Apparently, under the reaction conditions, metal-halogen exchange and  $\alpha$ -depro-



tonation are competitive processes, one leading to formation of the ethyl analogue **24** (after quenching and isolation) and the other yielding the desired bicyclo[2.2.1] product **2a**.

Although *n*-BuLi is known to add to the 1,2 double bond of pyridine,<sup>19</sup> the intramolecular cyclization of **21** to **2a** proceeded very rapidly at -78 °C, thus avoiding this potential addition. Under these conditions, metal– halogen exchange was not observed and **2a** was the only product, isolated in 55% yield. Treatment of **2a** with 3 M HCl in EtOAc removed the Boc group, and the amine **2b** was methylated (formic acid/formaldehyde)<sup>20</sup> to the target nicotine analogue **2**. Since **2** is a symmetric molecule it was unnecessary to carry out a parallel sequence from D-glutamic acid.

**Construction of the Bicyclo [3.2.1] Ring System.** From ester **15**, the intermediate needed for cyclization to the [3.2.1] system was constructed. Reduction of this ester and oxidation of the resulting alcohol **25** gave aldehyde **26**. Treatment with the sodium salt of trimethyl phosphono- acetate afforded olefin **27** as a mixture of the E and Z isomers (Scheme 5). This mixture of isomers was reduced to propionic ester **28**. Following the same procedure used for the formation of bromide **21**, ester **28** was reduced to the alcohol **29**, which was treated with the imidazolium mesylating reagent followed by LiBr to yield the desired bromide **31**.

Employing the same conditions used for formation of the [2.2.1] system, **31** was treated with *n*-BuLi at -78 °C in THF. Only a small amount of the desired [3.2.1] structure **3a** (<10%) was formed; the major products consisted of unidentified polar material. A possible explanation for this difference in behavior is that deprotonation at the  $\alpha$ -carbon to the pyridine in **31** did occur but that the intramolecular cyclization to the [3.2.1] system is less favorable than formation of the [2.2.1] system, thus permitting intermolecular side reactions to occur. To reduce the extent of any competing intermolecular reactions, higher dilution conditions were applied. Bromide **31** was added to the solution of *n*-Buli in THF via a syringe pump over a period of 1 h. Indeed, an improved yield of 48% was obtained. When the same technique was applied to bromide **21**, the yield of the [2.2.1] system product **2a** was improved from 55% to 64%. Deprotection and methylation gave the final [3.2.1] azabicyclo (–)-**3** analogue. To obtain the enantiomer (+)-**3**, exactly the same sequence of reactions was applied to D-glutamic acid.

#### Conclusion

We report a process for the enantiospecific synthesis of nicotine analogues in which the pyrrolidine ring is conformationally constrained into an azabicyclo ring system. The specific systems prepared are the 1-(3pyridinyl)-7-azabicyclo[2.2.1]heptanes and the 1-(3-pyridinyl)-8-azabicyclo[3.2.1] octanes. Enantiospecificity is achieved by basing the synthetic sequence on L- or D-glutamic acid to which the metallopyridine is attached. Conceivably, the method could be extrapolated to other substitution patterns and structural types.

## **Experimental Section**

**General Procedures.** All melting points are uncorrected. <sup>1</sup>H NMR and <sup>13</sup>C spectra were recorded in CDCl<sub>3</sub> unless otherwise specified, and chemical shifts are reported in ppm from internal tetramethylsilane; <sup>13</sup>C chemical shifts are reported in ppm from CDCl<sub>3</sub> or CD<sub>3</sub>OD, followed by multiplicity as determined from DEPT; *J* values are given in Hz. Reagents and solvents were dried and purified by distillation before use: CH<sub>2</sub>Cl<sub>2</sub> from CaH<sub>2</sub>; THF and Et<sub>2</sub>O from Na; MeOH from Mg. Reactions were conducted under N<sub>2</sub> and organic extracts were dried over Na<sub>2</sub>SO<sub>4</sub> prior to evaporation. Chromatography

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Scheme 5. Synthesis of the (+)- and (-)-[3.2.1] azabicyclonicotine Analogues



was carried out on gravity-grade  $SiO_2$ . Elemental analyses were determined by Microanalytical Laboratory, University of California at Berkeley.

tert-Butyl (2S)-2-(N-(tert-Butyloxycarbonyl)amino)-5oxo-5-(3-pyridinyl)pentanoate ((+)-8). Method A. To a solution of 3-iodopyridine (123 mg, 0.6 mmol) in THF (5 mL) was added EtMgBr (0.72 mL of a 1.0 M THF solution, 0.72 mmol) at room temperature, and the mixture was stirred for 25 min. To a solution of N-(tert-butyloxycarbonyl)-L-glutamic acid  $\alpha$ -*tert*-butyl ester (4, 152 mg, 0.5 mmol) in dichloromethane (2 mL) at -20 °C were added 2,3-dihydrofuran (38.5 mg, 0.55 mmol) and a solution of methanesulfonic acid in dichloromethane (5  $\mu$ L, 0.5 M solution, 2.5  $\mu$ mol). The mixture was allowed to warm to 0 °C over 4 h and then recooled to  $-20\,$  °C, and to this mixture was added the 3-pyridinyl Grignard reagent dropwise over 5 min to yield a clear, colored solution. After 30 min at -20 °C, the mixture was very slowly warmed to room temperature over 18 h and then recooled to 0 °C and added dropwise to aqueous 1 M K<sub>2</sub>-CO<sub>3</sub>. The mixture was extracted with dichloromethane (3  $\times$ 20 mL), and the combined organic layer was dried, filtered, and evaporated to a residue that was chromatographed (CH<sub>2</sub>-Cl<sub>2</sub>/EtOAc, 7/3) to afford 8 (55 mg, 30% yield) as a white solid: mp 89–90 °C;  $[\alpha]^{22}_{D}$  +8.7 (*c* 1, CHCl<sub>3</sub>); <sup>1</sup>H NMR  $\delta$  1.42 (s, 9H), 1.48 (s, 9H), 2.01-2.13 (m, 1H), 2.28-2.40 (m, 1H), 3.01-3.30 (m, 2H), 4.28 (m, 1H), 5.26 (d, J = 7.5, 1H), 7.43 (dd, J = 7.9, 4.8, 1H), 8.24 (dt, J = 7.9, 1.7, 1H), 8.79 (d, J = 4.8, 1H), 9.17 (s, 1H); <sup>13</sup>C NMR & 27.0, 27.9, 28.1, 34.7, 53.3, 79.7, 82.1, 123.5, 131.9, 135.2, 149.4, 153.4, 155.4, 171.3, 197.7. Anal. Calcd for C19H28N2O5: C, 62.6; H, 7.7; N, 7.7. Found: C, 62.8; H, 7.8; N, 7.6.

**Method B.** 3-Bromopyridine (0.241 mL, 2.5 mmol) in ether (3.5 mL) was added dropwise to a solution of *n*-BuLi (1.56 mL of a 1.6 M solution in hexane, 2.5 mmol) in ether (10 mL) at -78 °C in 5 min. The resulting yellow slurry was stirred at -78 °C for 30 min, and a solution of *N*-(*tert*-butoxycarbonyl)-L-glutamic acid  $\alpha$ -*tert*-butyl ester  $\beta$ -isoxazolidide (5, 354 mg, 1 mmol) in THF (5 mL) was added over 5 min. The resulting mixture was warmed to room temperature over 6 h, stirred at room temperature for 12 h, diluted with ether (30 mL), and quenched with H<sub>2</sub>O (40 mL). The aqueous layer was extracted with ether (3 × 30 mL), and the combined organic layer was dried, filtered, and evaporated to a residue, which was chromatographed (SiO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>/EtOAc, 7/3), yielding a mixture of **8** and **5** (85/15, 293 mg); 69% yield of **8**.

**Method C.** To a solution of isoxazolidide **5** (177 mg, 0.5 mmol) in THF (2 mL) at room temperature was added the 3-pyridinyl Grignard reagent (0.72 mmol) dropwise over 5 min to yield a clear, colored solution (0.72 mmol) that was stirred for 9 h at room temperature. Aqueous  $K_2CO_3$  was added, the mixture was extracted with dichloromethane (3 × 20 mL), and the combined organic layer was dried, filtered, and evaporated. Chromatography (CH<sub>2</sub>Cl<sub>2</sub>/EtOAc, 7/3) of the residue afforded a mixture of **8** and **5** (63/37, 143 mg); 50% yield of **8**.

**Method D.** 3-Bromopyridine (7.23 mL, 74.7 mmol) in Et<sub>2</sub>O (57 mL) was added dropwise to a solution of *n*-BuLi (1.6 M solution in hexane, 47 mL, 74.7 mmol) in Et<sub>2</sub>O (167 mL) at -78 °C over 30 min. The resulting yellow slurry was stirred at -78 °C for 30 min, and a solution of *tert*-butyl *N*-(*tert*-butyloxycarbonyl)-L-pyroglutamate (**6**, 14.3 g, 50 mmol) in Et<sub>2</sub>O/THF (70 mL, 1/1) was added over 10 min. The resulting mixture was stirred at -78 °C for 1 h, poured into H<sub>2</sub>O (60 mL), and extracted with Et<sub>2</sub>O (50 mL). The combined organic layer was dried, filtered, and evaporated, and the residue was chromatographed (CH<sub>2</sub>Cl<sub>2</sub>/EtOAc, 7/3) to afford **8** (13.1 g, 72%).

Methyl (2S)-2-(N-(tert-Butyloxycarbonyl)amino)-5-oxo-5-(3-pyridinyl)pentanoate ((+)-9). 3-Bromopyridine (7.23 mL, 74.7 mmol) in Et<sub>2</sub>O (57 mL) was added dropwise to a solution of n-BuLi (1.6 M solution in hexane, 47 mL, 74.7 mmol) in Et\_2O (167 mL) at  $-78\ ^\circ C$  over 30 min. The resulting yellow slurry was stirred at -78 °C for 30 min, and a solution of methyl N-(tert-butyloxycarbonyl)-L-pyroglutamate (7, 16.4 g, 67.3 mmol) in THF/Et<sub>2</sub>O (80 mL, 1/1) was added over 20 min. The resulting mixture was stirred at -78 °C for 1 h and then poured into  $H_2O$  (60 mL), the aqueous layer was extracted with Et<sub>2</sub>O (50 mL), and the combined organic layer was dried, filtered, and evaporated. The residue was chromatographed  $(CH_2Cl_2/EtOAc, 7/3)$  to afford **9** (12.9 g, 59%) as a white solid: mp 90–92 °C;  $[\alpha]^{22}_{D}$  +13.5 (c 1, CHCl<sub>3</sub>); <sup>1</sup>H NMR  $\delta$  1.40 (s, 9H), 2.06 (m, 1H), 2.34 (m, 1H), 3.01-3.19 (m, 2H), 3.75 (s, 3H), 4.40 (m, 1H), 5.22 (d, J = 7.5, 1H), 7.41 (dd, J = 7.7, 4.9, 1H), 8.22 (dt, J = 7.7, 1.7, 1H), 8.77 (d, J = 4.9, 1H), 9.15 (d, J = 1.7, 1H); <sup>13</sup>C NMR  $\delta$  26.6, 28.1, 34.6, 52.3, 52.7, 80.0, 123.5,

131.8, 135.2, 149.4, 153.4, 155.3, 172.6, 197.5. Anal. Calcd for  $C_{16}H_{22}N_2O_5\!\!:$  C, 59.6; H, 6.9; N, 8.7. Found: C, 59.7; H, 6.9; N, 8.6.

Methyl (2*R*)-2-(*N*-(*tert*-butyloxycarbonyl)amino)-5-oxo-5(3-pyridinyl)pentanoate ((-)-9, structure not shown) was prepared from the corresponding D-pyroglutamate: mp 91– 93 °C;  $[\alpha]^{22}_D$  –14.1 (*c* 1, CHCl<sub>3</sub>); spectral and chromatographic properties identical with those of (+)-9.

Methyl (2.5)-Amino-5-oxo-5-(3-pyridinyl)pentanoate Dihydrochloride (10), (2S)-2-(3-Pyridinyl)-5-methoxycarbonyl-1-pyrroline (12), and (2S)-cis-5-(3-Pyridinyl)proline Methyl Ester ((+)-13). To pyridinyl ketone (+)-9 (2.65 g, 8.24 mmol) was added slowly, while stirring, HCl in EtOAc (3 M, 30 mL) at room temperature. The mixture was stirred overnight while precipitate formed; concentration under reduced pressure gave crude dihydrochloride 10 as a pale yellow solid: <sup>1</sup>H NMR  $\delta$  2.25–2.50 (m, 2H), 3.41 (q, J = 6.6, 2H), 3.79 (s, 3H), 4.26 (t, J = 6.9, 1H), 8.22 (t, J = 7.0, 1H), 8.98 (d, J = 4.2, 1H), 9.09 (d, J = 8.2, 1H), 9.33 (s, 1H). The crude **10** was dissolved in H<sub>2</sub>O (20 mL) and shaken with EtOAc (100 mL) and saturated NaHCO<sub>3</sub> in H<sub>2</sub>O (20 mL). The aqueous layer was extracted with EtOAc ( $2 \times 60$  mL), and the combined organic layer was dried, filtered, and evaporated to afford crude pyrroline **12** (1.46 g, 87%) as a yellow oil: <sup>1</sup>H NMR  $\delta$ 2.15-2.40 (m, 2H), 2.85-3.05 (m, 1H), 3.08-3.20 (m, 1H), 3.74 (s, 3H), 4.89 (dd, J = 8.7, 6.6, 1H), 7.31 (dd, J = 7.9, 4.8, 1H), 8.19 (dt, J = 7.9, 1.8, 1H), 8.62 (dd, J = 4.8, 1.6, 1H), 8.97 (d, J = 2.0, 1H); <sup>13</sup>C NMR  $\delta$  26.1 (CH<sub>2</sub>), 35.2 (CH<sub>2</sub>), 52.2 (CH<sub>3</sub>), 74.5 (CH), 123.3 (CH), 129.4 (C), 135.0 (CH), 149.2 (CH), 151.7 (CH), 172.9 (C), 173.8 (C). Pyrroline 12 (1.2 g, 5.88 mmol) in IPA (15 mL) was treated with  $H_2$  at 1 atm in the presence of 10% Pd/C (0.12 g, 10% w/w) at room temperature overnight. The resulting mixture was filtered and evaporated to give 13 (1.2 g, 95%) as a colorless oil. This crude product was suitable for further reactions. An analytic sample was obtained by chromatography (CH<sub>2</sub>Cl<sub>2</sub>/MeOH saturated with NH<sub>3</sub>, 95/5):  $[\alpha]^{22}_{D}$  +34.2 (c 1.1, CHCl<sub>3</sub>); <sup>1</sup>H NMR  $\delta$  1.58–1.71 (m, 1H), 2.01-2.25 (m, 3H), 2.41 (bs, 1H), 3.70 (s, 3H), 3.90 (dd, J =1H), 7.80 (dt, J = 7.8, 1.8, 1H), 8.43 (dd, J = 4.8, 2.5, 1H), 8.54 (d, J = 2.0, 1H); <sup>13</sup>C NMR  $\delta$  30.0 (CH<sub>2</sub>), 34.0 (CH<sub>2</sub>), 51.9 (CH<sub>3</sub>), 59.7 (CH), 60.6 (CH), 123.3 (CH), 134.1 (CH), 138.9 (C), 148.4 (CH), 148.6 (CH), 175.1 (C). Anal. Calcd for C<sub>11</sub>H<sub>14</sub>N<sub>2</sub>O<sub>2</sub>: C, 64.1; H, 6.8; N, 13.6. Found: C, 63.7; H, 6.9; N, 13.5.

(2*R*)-*cis*-5-(3-Pyridinyl)proline methyl ester ((–)-13, structure not shown) was prepared from the corresponding pyridinyl ketone (–)-9:  $[\alpha]_D 22-35.0$  (*c* 1.3, CHCl<sub>3</sub>); spectral and chromatographic properties identical with those of (+)-13.

Methyl (2S)-(N-(tert-Butyloxycarbonyl)amino)-5,5dimethoxy-5-(3-pyridinyl)pentanoate (11). Pyridinyl ketone 8 (0.3 g, 0.824 mmol) was treated with saturated HCl in MeOH (10 mL) at room temperature for 4 h. The mixture was evaporated, and the residue was dissolved in H<sub>2</sub>O (5 mL) and distributed between EtOAc (20 mL) and saturated NaHCO<sub>3</sub> in H<sub>2</sub>O (10 mL). The aqueous layer was extracted with EtOAc  $(2 \times 20 \text{ mL})$ , and the combined organic layer was dried, filtered, and evaporated. The residue was chromatographed (CH<sub>2</sub>Cl<sub>2</sub>/MeOH saturated with NH<sub>3</sub>, 97/3) to afford first 12 (126 mg, 75%) then 11 (38 mg, 18%) both as colorless oils. For 11: <sup>1</sup>H NMR δ 1.05–1.45 (m, 2H), 1.92–2.15 (m, 2H), 3.14 (s, 3H), 3.15 (s,3H), 3.63 (s, 3H), 7.27 (dd, J = 7.9, 4.8, 1H), 7.73 (dt, J = 7.9, 2.0, 1H), 8.53 (dd, J = 4.8, 1.7, 1H), 8.67 (d, J = 2.2, 1.51H); <sup>13</sup>C NMR & 28.5, 33.1, 48.6, 51.8, 53.9, 102.2, 122.9, 134.6, 135.8, 148.7, 149.0, 175.9. Treatment of 11 with aqueous HCl at pH 1.5 for 2 h, concentration of the mixture, and partition between saturated NaHCO3 in H2O and EtOAc as above, led to complete conversion to 12.

(2.5)-trans-5-(3-Pyridinyl)proline Methyl Ester (14). Pyrroline 12 (300 mg, 1.47 mmol) in MeOH/Acetate buffer (25 mL, 1/1, v/v) was treated with NaCNBH<sub>3</sub> (110 mg, 1.76 mmol) at room temperature overnight. The mixture was poured into saturated NaHCO<sub>3</sub> in H<sub>2</sub>O (25 mL) and extracted with CHCl<sub>3</sub>/ IPA (3/1, v/v,  $3 \times 30$  mL). The combined organic layer was dried, filtered, and evaporated. The residue was chromatographed (CH<sub>2</sub>Cl<sub>2</sub>/MeOH saturated with NH<sub>3</sub>, 95/5) to afford first **14** (121 mg, 40%) and then **13** (153 mg, 51%), both as colorless oils. For **14**: <sup>1</sup>H NMR  $\delta$  1.60–1.80 (m, 1H), 1.90– 2.05 (m, 1H), 2.05–2.28 (m, 1H), 2.30–2.38 (m, 1H), 2.64 (br, 1H), 3.75 (s, 3H), 4.03 (dd, J = 8.4, 5.7, 1H), 4.39 (t, J = 7.2, 1H), 7.23 (dd, J = 7.8, 4.7, 1H), 7.80 (dt, J = 7.8, 1.7, 1H), 8.47 (dd, J = 4.7, 1.4, 1H), 8.59 (d, J = 1.9, 1H); <sup>13</sup>C NMR  $\delta$ 29.7 (CH<sub>2</sub>), 34.6 (CH<sub>2</sub>), 52.2 (CH<sub>3</sub>), 59.2 (CH), 59.4 (CH), 123.3 (CH), 134.1 (CH), 139.8 (C), 148.4 (CH), 148.5 (CH), 176.2 (C). Anal. Calcd for C<sub>11</sub>H<sub>14</sub>N<sub>2</sub>O<sub>2</sub>·<sup>1</sup>/<sub>4</sub>H<sub>2</sub>O: C, 62.7; H, 6.9; N, 13.3. Found: C,62.9; H,7.1; N, 13.3.

(2S)-cis-1-(tert-Butyloxycarbonyl)-5-(3-pyridinyl)proline Methyl Ester ((+)-15). Proline (+)-13 (3.4 g, 16.5 mmol) in THF (160 mL) at room temperature was treated with Et<sub>3</sub>N (2.5 g, 24.8 mmol) followed by (BOC)<sub>2</sub>O (4.32 mg, 19.8 mmol). The mixture was stirred at room temperature for 1 h and evaporated, and the residue was distributed between EtOAc (100 mL) and saturated NaHCO<sub>3</sub> in H<sub>2</sub>O (50 mL). The aqueous layer was extracted with EtOAc ( $2 \times 50$  mL), the combined organic layer was dried, filtered, and evaporated, and the residue was chromatographed (CH2Cl2/EtOAc, 4/6) to afford **15** (4.9 g, 97%) as a colorless oil:  $[\alpha]^{22}_{D}$  +27.1 (*c* 1.2, CHCl<sub>3</sub>); <sup>1</sup>H NMR (298 K, rotamer ratio, 60/40, \*denotes minor rotamer)  $\delta$  1.16 (s, 5.4H), 1.41\* (s, 3.6H), 1.90–2.18 (m, 2H), 2.20–2.30 (m, 1H), 2.32-2.42 (m, 1H), 3.82 (s, 3H),  $4.39^*$  (t, J = 8.4, 0.4H), 4.51-4.58 (m, 0.6H), 4.78 (t, J = 7.4, 0.6H),  $4.98-5.03^{*}$ (m, 0.4H), 7.20-7.40 (m, 1H),  $8.01^*$  (d, J = 7.7, 0.4H), 8.15 (d, J = 7.7, 0.6H, 8.50 (br, 1H), 8.60 (bs, 0.6H), 8.69\* (bs, 0.4H); <sup>1</sup>H NMR (C<sub>6</sub>D<sub>6</sub>, 298 K, rotamer ratio 70/30)  $\delta$  1.09 (s, 6.3H), 1.32\* (s, 2.7H), 1.40-1.60 (m, 4H), 3.30 (s, 2.1H), 3.34\* (s, 0.9H), 4.15\* (br, 0.3H), 4.28 (br, 0.7H), 4.41 (br, 0.7H), 4.72\* (br, 0.3H), 6.75-6.85 (m, 1H),  $8.00^*$  (bd, J = 7.8, 0.3H), 8.21(d, J = 7.8, 0.7H), 8.42 (br, 1H), 8.70 (bs, 0.7H), 8.84\* (bs, 0.3H); <sup>1</sup>H NMR (C<sub>6</sub>D<sub>6</sub>, 343 K)  $\delta$  1.20 (s, 9H), 1.45–1.58 (m, 4H), 3.38 (s, 3H), 4.33 (br, 1H), 4.53 (br, 1H), 6.86 (dd, J =7.8, 4.7, 1H), 7.97 (d, J = 7.8, 1H), 8.41 (dd, J = 4.7, 1.6, 1H), 8.73 (s, 1H); <sup>13</sup>C NMR very complex because of rotamers. Anal. Calcd for C<sub>16</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>: C, 62.7; H, 7.2; N, 9.1. Found: C, 62.4; H, 7.1; N, 9.2.

(2*R*)-*cis*-1-(*tert*-Butyloxycarbonyl)-5-(3-pyridinyl)proline methyl ester ((–)-15, structure not shown) was prepared from the corresponding proline (–)-13:  $[\alpha]^{22}_{D}$  –27.3 (*c* 1.1, CHCl<sub>3</sub>); spectral and chromatographic properties identical with those of (+)-15.

(2S)-cis-1-(tert-Butyloxycarbonyl)-5-(3-pyridinyl)proline (16), (2S,5R)-1-(tert-Butyloxy-carbonyl)-2-(1-diazomethylcarbonyl)-5-(3-pyridinyl)pyrrolidine (17), and (2S,5R)-1-(tert-Butyl-oxycarbonyl)-2-(methoxycarbonylmethyl)-5-(3-pyridinyl)pyrrolidine (18). To methyl ester 15 (3.0 g, 9.8 mmol) in MeOH/H<sub>2</sub>O (3/1, v/v, 100 mL) at room temperature was added LiOH·H<sub>2</sub>O (1.65 g, 39.2 mmol). The solution was stirred at room temperature for 5 h, poured into aqueous H<sub>3</sub>PO<sub>4</sub> solution (0.1 M, 150 mL), and extracted with CHCl<sub>3</sub>/IPA (3/1, v/v,  $3 \times 150$  mL). The combined organic layer was dried, filtered, and evaporated to afford acid 16 (2.8 g, 97%) as a white solid that is suitable for the next reaction without further purification: <sup>1</sup>H NMR (298 K, rotamer ratio 60/40)  $\delta$  1.14 (s, 9H), 1.36 (s, 9H), 1.80–2.40 (m, 4H), 4.39 (t, J = 7.4, 0.6H),  $4.56^*$  (dd, J = 7.0, 5.0, 0.4H),  $4.81^*$  (t, J = 7.5, 0.4H), 5.04 (dd, J = 7.5, 4.2, 0.6H), 7.38\* (m, 0.4H), 8.42 (dd, J = 7.8, 5.3, 0.6H), 7.82 (d, J = 7.8, 0.6H), 7.86\* (d, J = 7.7, 0.6H) 0.4H), 8.59 (br, 1H), 9.19\* (bs, 0.4H), 9.22 (bs, 0.6H). To acid **16** (0.93 g, 3.17 mmol) in THF (15 mL) at -15 °C was added Et<sub>3</sub>N (440  $\mu$ L, 3.17 mmol) followed by ethyl chloroformate (300  $\mu$ L, 3.17 mmol). The mixture was stirred at -15 °C for 15 min then warmed to 0 °C followed by addition of excess CH<sub>2</sub>N<sub>2</sub> (0.5 M in Et<sub>2</sub>O, 20 mL, 10.0 mmol). The reaction mixture stayed bright yellow and was stirred for 5 h, warmed to room temperature slowly, and stirred vigorously until the bright yellow color disappeared. The mixture was evaporated, and the residue was distributed between EtOAc (50 mL) and saturated aqueous NaHCO<sub>3</sub> (25 mL). The aqueous layer was

extracted with EtOAc ( $2 \times 25$  mL), and the combined organic layer was dried, filtered, and evaporated, affording crude diazo ketone 17 (0.95 g) as a pale yellow oil that is suitable for the next reaction without further purification: <sup>1</sup>H NMR (298 K, rotamers)  $\delta$  1.00–1.50 (br, 9H), 1.90–2.40 (m, 4H), 4.45 (br, 1H), 4.60-5.10 (br, 1H), 5.35-5.75 (br, 1H), 7.29 (dd, J = 7.8, 4.7, 1H), 8.02 (bd, J = 7.8, 1H), 8.49 (dd, J = 4.7, 2.0, 1H), 8.55 (br, 1H); <sup>1</sup>H NMR (323 K)  $\delta$  1.27 (s, 9H), 1.95–2.05 (m, 2H), 2.20-2.40 (m, 2H), 4.84 (br, 1H), 4.80 (br, 1H), 5.58 (bs, 1H), 7.27 (dd, J = 7.9, 4.7, 1H), 7.96 (d, J = 7.9, 1H), 8.50 (d, J = 4.7, 1H, 8.58 (s, 1H). To diazoketone **17** (0.95 g, 3 mmol) in MeOH (12 mL) at -25 °C was added a solution of PhCO2-Ag (80 mg, 0.33 mmol) in  $Et_3N$  (880 mg, 8.7 mmol) with the exclusion of light. The mixture was warmed to 0 °C over 3 h and was stirred overnight while it warmed to room temperature. The reaction mixture was evaporated, the residue was distributed between EtOAc (50 mL) and saturated aqueous NaHCO<sub>3</sub> (25 mL), the aqueous layer was extracted with EtOAc (2  $\times$  20 mL), and the combined organic layer was dried and filtered. The filtrate was evaporated and the residue chromatographed (CH<sub>2</sub>Cl<sub>2</sub>/EtOAc, 4/6) to afford homologous ester 18 (700 mg, 70% from 15) as a colorless oil: <sup>1</sup>H NMR (298 K, rotamers) & 1.00-1.60 (br, 9H), 1.70-1.95 (m, 2H), 1.90-2.18 (m, 2H), 2.08-2.25 (m, 1H), 2.25-2.41 (m, 1H), 2.48 (dd, J= 14.8, 9.5, 1H), 3.08 (br, 1H), 3.90 (s, 3H), 4.40 (br, 1H), 4.65-5.00 (br, 1H), 7.27 (br, 1H), 7.56 (bd, J = 7.8, 1H), 8.50 (br, 2H); <sup>1</sup>H NMR (C<sub>6</sub>D<sub>6</sub>, 298 K, rotamers)  $\delta$  1.14 (br, 9H), 1.10-1.40 (m, 2H), 1.60–1.80 (m, 2H), 2.26 (dd, J = 14.8, 9.3, 1H), 3.02 (br, 1H), 3.36 (s, 3H), 4.28 (br, 1H), 4.41 (br, 1H), 6.82 (dd, J = 4.9, 6.7, 1H), 7.29 (dt, J = 7.9, 1.8, 1H), 8.45 (d, J =4.7, 1H), 9.11 (br, 1H); <sup>1</sup>H NMR (C<sub>6</sub>D<sub>6</sub>, 329 K)  $\delta$  1.21 (s, 9H), 1.35.1.50 (m, 2H), 1.60–1.80 (m, 2H), 2.30 (dd, J = 14.7, 9.2, 1H), 3.03 (dd, J = 14.7, 4.4, 1H), 3.39 (s, 3H), 4.35 (br, 1H), 4.50 (br, 1H), 6.84 (dd, J = 7.8, 4.8, 1H), 7.31 (d, J = 7.8, 1H), 8.43 (d, J = 4.8, 1H), 8.57 (s, 1H); <sup>13</sup>C NMR (C<sub>6</sub>D<sub>6</sub>, 298 K, rotamers)  $\delta$  28.2, 30.2 (br), 33.8 (br), 40.1 (br), 51.2, 56.2, 60.8, 79.5, 123.2, 132.6, 140.0 (br), 148.6, 148.7, 154.3, 171.3; <sup>13</sup>C NMR (C<sub>6</sub>D<sub>6</sub>, 328 K) & 28.2 (CH<sub>3</sub>), 30.2 (CH<sub>2</sub>), 33.8 (CH<sub>2</sub>), 40.1 (CH<sub>2</sub>), 51.0 (CH<sub>3</sub>), 56.3 (CH), 60.9 (CH), 79.5 (C), 123.0 (CH), 132.7 (CH), 140.0 (C), 148.5 (CH), 148.7 (CH), 154.4 (C), 171.2 (C). Anal. Calcd for C<sub>17</sub>H<sub>24</sub>N<sub>2</sub>O<sub>4</sub>: C, 63.7; H, 7.5; N, 8.7; Found: C, 63.6; H, 7.5; N, 8.7.

(2S,5R)-1-(tert-Butyloxycarbonyl)-2-(2-hydroxyethyl)-5-(3-pyridinyl)pyrrolidine (19). To a solution of ester 18 (2.0 g, 6.25 mmol) in EtOH (200 mL) at 0 °C was added CaCl<sub>2</sub> (1.47 g, 12.5 mmol) followed by NaBH<sub>4</sub> (950 mg, 25 mmol). The mixture was stirred overnight while it warmed to room temperature, and then aqueous K<sub>2</sub>CO<sub>3</sub> (2M, 50 mL) was added and the mixture was evaporated. The residue was distributed between EtOAc (100 mL) and H<sub>2</sub>O (50 mL), the aqueous layer was extracted with EtOAc (2  $\times$  40 mL), and the combined organic layer was dried, filtered, and evaporated. The residue was chromatographed (CH<sub>2</sub>Cl<sub>2</sub>/EtOAc 2/8) to afford alcohol 19 (1.3 g, 71%) as a colorless oil:  $[\alpha]^{22}_{D}$  +39.6 (c 1, CHCl<sub>3</sub>); <sup>1</sup>H NMR (298 K, rotamer ratio, 60/40)  $\delta$  1.22 (bs, 5.4H), 1.25\* (bs, 3.6H), 1.60-2.00 (m, 4H), 2.00-2.20 (m, 1H), 2.25-2.40 (m, 1H), 3.80 (bm, 2H), 4.19 (br, 0.6H), 4.38 (br, 0.6H), 4.49\* (br, 0.4H),  $4.55-4.67^*$  (br, 0.4H), 4.71 (bt, J = 7.6, 0.6H),  $4.86^*$ (bt, J = 7.1, 0.4H), 7.25 (br, 1H), 7.51 (d, J = 7.9, 0.6H), 7.61\* (d, J = 7.9, 0.4H), 8.50 (br, 2H); <sup>1</sup>H NMR (C<sub>6</sub>D<sub>6</sub>, 298 K, rotamers) & 1.07 (bs, 9H), 1.20-1.70 (bm, 4H), 3.60-3.85 (br, 2H), 4.21 (br, 2H), 4.66 (br, 1H), 6.75 (dd, J = 7.9, 4.7, 1H), 7.13 (br, 1H), 8.42 (bd, J = 4.7, 1H), 8.51 (br, 1H); <sup>1</sup>H NMR  $(C_6D_6, 329 \text{ K}) \delta 1.22 \text{ (s, 9H)}, 1.25 - 1.38 \text{ (m, 1H)}, 1.40 - 1.72 \text{ (m, })$ 4H), 1.75-1.83 (m, 1H), 3.70-3.90 (m, 2H), 3.80-4.20 (br, 1H), 4.28 (bq, J = 7.0, 1H), 4.45 (dt, J = 7.4, 1H), 6.89 (bd, J = 7.9, 4.7, 1H), 7.31 (d, J = 7.9, 1H), 8.51 (d, J = 4.7, 1H), 8.63 (s, 1H); <sup>13</sup>C NMR (C<sub>6</sub>D<sub>6</sub>, 298 K)  $\delta$  28.0 (CH<sub>3</sub>), 30.0 (CH<sub>2</sub>), 34.8 (CH<sub>2</sub>), 39.2 (CH<sub>2</sub>), 55.9 (CH), 59.2 (CH<sub>2</sub>), 61.3 (CH), 80.1 (C), 123.2 (CH), 132.6 (CH), 140.1 (C), 148.6 (CH), 148.8 (CH), 156.3 (C); <sup>13</sup>C NMR (C<sub>6</sub>D<sub>6</sub>, 329 K)  $\delta$  27.7, 30.4, 34.2, 39.1, 55.9, 59.1, 61.0, 79.7, 122.7, 132.2, 139.7, 148.2, 148.4, 155.7. Anal. Calcd for  $C_{16}H_{24}N_2O_3:\,$  C, 65.7; H, 8.3; N, 9.6. Found: C, 65.5; H, 8.2; N, 9.6.

(2S.5R)-1-(tert-Butyloxycarbonyl)-2-(2-methylsulfonyloxyethyl)-5-(3-pyridinyl)pyrrolidine (20), (2S,5R)-1-(tert-Butyloxycarbonyl)-2-(2-bromoethyl)-5-(3-pyridinyl)pyrrolidine (21), and Cyclic Carbamate 22. To a solution 1-methanesulfonylimidazolide (180 mg, 1.2 mmol) in THF (12 mL) at 0 °C was added methyl triflate (127 µL, 1.2 mmol) dropwise, the mixture was stirred at 0  $^\circ C$  for 30 min, and then a mixture of alcohol 19 (300 mg, 1.03 mmol) and 1-methylimidazole (84 µL, 1.03 mmol) in THF (3 mL) was added dropwise at 0 °C. After addition, the ice bath was removed, the mixture was stirred overnight and evaporated, and the residue was dissolved in EtOAc (10 mL) and washed with H<sub>2</sub>O (5 mL). The organic layer was dried and evaporated to afford crude mesylate 20 (containing a small amount of 1-mesylimidazolide): <sup>1</sup>H NMR (298 K, rotamers)  $\delta$  1.00–1.50 (br, 9H), 1.75-(br, 1H), 1.80-2.00 (m, 2H), 2.13 (m, 1H), 2.27-2.44 (m, 2H), 3.00 (s, 3H), 4.12 (bs, 1H), 4.37 (bs, 2H), 4.60-4.80 (br, 1H), 7.25 (br, 1H), 7.55 (bd, J = 7.9, 1H), 8.51 (bs, 2H). Crude mesylate 20 was dissolved in acetone (12 mL), LiBr (358 mg, 4.12 mmol) was added, and the mixture was heated under reflux for about 1 h, with monitoring by TLC. After evaporation, the residue was distributed between CH<sub>2</sub>Cl<sub>2</sub> (10 mL) and saturated aqueous NaHCO<sub>3</sub> (5 mL), the aqueous layer was extracted with  $CH_2Cl_2$  (2  $\times$  10), and the combined organic layer was dried, filtered, and evaporated. The residue was chromatographed (CH<sub>2</sub>Cl<sub>2</sub>/EtOAc 4/6) to afford bromide 21 (260 mg, 73%) as a colorless oil; elution with EtOAc gave cyclic carbamate 22 (23 mg, 11% yield) as a white solid with a broad (60–80 °C) mp. For bromide **21**: <sup>1</sup>H NMR (298 K, rotamers)  $\delta$ 1.00-1.60 (br, 9H), 1.65-1.79 (br, 1H), 1.80-1.95 (m, 1H), 1.91-2.20 (m, 2H), 2.28-2.40 (m, 1H), 2.43-2.60 (br, 1H), 3.37-3.60 (m, 2H), 4.10-4.20 (m, 1H), 4.60-5.00 (br, 1H), 7.27 (dd, J = 7.9, 4.9, 1H), 7.52 (dt, J = 7.9, 1.8, 1H), 8.51 (bs, 2H);<sup>1</sup>H NMR (C<sub>6</sub>D<sub>6</sub>, 298 K, rotamers) δ 0.90–1.80 (br, 14H), 2.20– 2.40 (br, 1H), 2.95-3.25 (br, 2H), 3.89 (br, 1H), 4.20-4.80 (br, 1H), 6.74 (dd, J = 7.8, 4.6, 1H), 7.15 (overlap with solvent, 1H), 8.47 (d, J = 4.6, 1H), 8.59 (br, 1H); <sup>1</sup>H NMR (C<sub>6</sub>D<sub>6</sub>, 333 K)  $\delta$  1.05–1.20 (m, 1H), 1.24 (s, 9H), 1.15–1.28 (m, 2H), 1.60– 1.80 (m, 2H), 2.38 (p, J = 6.9, 1H), 3.10–3.29 (m, 2H), 3.91 (m, 1H), 4.50 (dt, J = 7.4, 1H), 6.79 (dd, J = 7.7, 4.6, 1H), 7.21 (bd, J = 7.7, 1H), 8.44 (d, J = 4.6, 1H), 8.58 (s, 1H); <sup>13</sup>C NMR (C<sub>6</sub>D<sub>6</sub>, 298 K) & 28.3 (CH<sub>3</sub>), 29.8 (CH<sub>2</sub>), 30.0 (CH<sub>2</sub>), 33.8 (CH<sub>2</sub>), 39.6 (CH<sub>2</sub>), 58.4 (CH), 60.9 (CH), 79.7 (C), 123.0 (CH), 132.7 (CH), 139.7 (C), 148.6 (CH), 148.7 (CH), 154.9 (C). For cyclic carbamate 22: <sup>1</sup>H NMR & 1.70–1.86 (m, 2H), 1.93–2.04 (m, 1H), 2.12 (m, 1H), 2.20–2.40 (m, 2H), 3.75 (ddt, J = 3.0, 5.1, 11.1, 1H), 4.28 (ddd, J = 3.3, 11.3, 12.8, 1H), 4.51 (ddd, J =1.4, 5.1, 11.3), 5.05 (d, J = 8.9, 1H), 7.24 (dd, J = 4.9, 7.9, 1H), 7.46 (dt, J = 1.9, 7.9, 1H), 8.48 (bs, 2H); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 28.4 (CH<sub>2</sub>), 29.9 (CH<sub>2</sub>), 32.6 (CH<sub>2</sub>), 57.4 (CH), 59.4 (CH), 67.4 (CH<sub>2</sub>), 123.2 (CH), 133.1 (CH), 138.1 (C), 147.3 (CH), 148.3 (CH), 155.7 (C). Anal. Calcd for C<sub>12</sub>H<sub>14</sub>N<sub>2</sub>O<sub>2</sub>: C, 66.0; H, 6.5; N, 12.8. Found: C, 66.1; H, 6.5; N, 12.7.

(2S,5R)-1-(tert-Butyloxycarbonyl)-2-vinyl-5-(3-pyridinyl)pyrrolidine (23). Methyltriphenylphosphonium bromide (2.64 g, 7.38 mmol) was suspended in 30 mL of THF, and with stirrring at 25 °C, a solution of n-BuLi (2.5 M in hexane, 2.95 mL, 7.38 mmol) was added dropwise over 15 min and the mixture was stirred for 3 h at 25 °C. An aliquot of this suspension (1.0 mL, 0.22 mmol) was placed in another flask, to it a solution of aldehyde 26 (0.18 mmol) in THF was added dropwise at 0 °C, and the resulting mixture stirred overnight. To the mixture, after being stirred overnight, was added pH 7 phosphate buffer (1 mL) followed by evaporation. The residue was distributed between EtOAc (10 mL) and H<sub>2</sub>O (5 mL), the organic layer was dried, filtered, and evaporated, and the residue was chromatographed (CH<sub>2</sub>Cl<sub>2</sub>/EtOAc 4/6) to afford vinyl pyrrolidine **23** (46 mg, 91%) as a colorless oil:  $[\alpha]^{22}_{D}$ +52.0 (c 1.3, CHCl<sub>3</sub>); <sup>1</sup>H NMR (298 K, rotamers)  $\delta$  1.00–1.60 (br, 9H), 1.85-1.95 (m, 2H), 2.09 (m, 1H), 2.28 (m, 1H), 4.20-4.50 (br, 1H), 4.60-4.95 (br, 1H), 5.17 (bd, 1H), 5.26 (bd, 1H), 5.95 (br, 2H), 7.22 (dd, J = 7.9, 4.9, 1H), 7.56 (dt, J = 7.9, 1.8, 1H), 8.42 (dd, J = 4.7, 1.4, 1H), 8.54 (d, J = 1.9, 1H); <sup>1</sup>H NMR (328 K)  $\delta$  1.27 (s, 9H), 1.75–1.95 (m, 2H), 2.07 (m, 1H), 2.27 (m, 1H), 4.45 (bs, 1H), 4.81 (bt, J = 6.8, 1H), 5.14 (d, J = 10.3, 1H), 5.23 (d, J = 17.1, 1H), 5.94 (m, 1H), 7.17 (dd, J = 7.9, 4.9, 1H), 7.53 (dt, J = 7.9, 1.8, 1H), 8.42 (d, J = 4.7, 1H), 8.51 (s, 1H); <sup>13</sup>C NMR (328 K)  $\delta$  28.2, 30.4, 34.0, 60.7, 60.8, 79.8, 115.1, 122.9, 133.3, 139.1, 139.4, 148.0, 148.2, 154.6. Anal. Calcd for C<sub>16</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>: C, 70.0; H, 8.1; N, 10.2. Found: C, 70.0; H, 8.1; N, 10.1.

(2S,5R)-1-(tert-Butyloxycarbonyl)-2-ethyl-5-(3-pyridinyl)pyrrolidine (24). Vinyl pyrrolidine 23 (20 mg, 0.073 mmol) in IPA (1 mL) was treated with H<sub>2</sub> at 1 atm in the presence of 10% Pd/C (2 mg, 10% w/w) at room temperature overnight. The resulting mixture was filtered and evaporated to give ethyl pyrrolidine 24 (20 g, 99%) as a colorless oil:  $[\alpha]^{22}_{D}$  +40.5 (*c* 1, CHCl<sub>3</sub>); <sup>1</sup>H NMR (298 K, rotamers)  $\delta$  0.95 (t, J = 7.4, 3H), 1.00-1.60 (bm, 10H), 1.69 (br, 1H), 1.81 (m, 1H), 1.98 (m, 2H), 2.29 (m, 1H), 3.86 (br, 1H), 4.69 (br, 1H), 7.20 (dd, J = 7.9, 4.9, 1H), 7.51 (dt, J = 7.9, 1.8, 1H), 8.44 (d, J =4.7, 1H), 8.48 (s, 1H); <sup>1</sup>H NMR (328 K)  $\delta$  0.96 (t, J = 7.4, 3H), 1.28 (s, 9H), 1.46 (m, 1H), 1.69 (m, 1H), 1.85 (m, 1H), 1.98 (m, 2H), 2.29 (m, 1H), 3.89 (m, 1H), 4.78 (t, J = 7.4,1H), 7.19 (dd, J = 7.9, 4.9, 1H), 7.52 (dt, J = 7.9, 1.8, 1H), 8.44 (d, J = 4.7, 1.8, 1H), 8.44 (d, J = 4.8, 1H), 8.44 (d, J = 4.8, 1H), 8.44 (d, J = 4.8, 1H), 8 1H), 8.50 (d, J = 1.8, 1H); <sup>13</sup>C NMR (328 K)  $\delta$  11.0, 28.3, 28.5, 29.3, 34.1, 60.7, 60.8, 79.6, 123.0, 133.1, 140.1, 148.0, 148.1, 155.0. Anal. Calcd for C<sub>16</sub>H<sub>24</sub>N<sub>2</sub>O<sub>2</sub>: C, 69.5; H, 8.7; N, 10.1. Found: C, 69.1; H, 8.7; N, 10.1.

7-(tert-Butyloxycarbonyl)-1-(3-pyridinyl)-7-azabicyclo-[2.2.1]heptane (2a). To a solution of bromide 21 (90 mg, 0.25 mmol) in THF (5 mL) at -78 °C was added n-BuLi (1.6 M in hexane, 0.55 mL) dropwise. The mixture was stirred at -78°C for 30 min, and then pH 7 buffer (10 mL) was added. The mixture was allowed to warm to room temperature and extracted with EtOAc ( $2 \times 20$  mL), and the combined organic layer was dried, filtered, and evaporated. The residue was chromatographed (hexanes/EtOAc, 1/1) to afford 2a (45 mg, 64%) as a white solid: mp 83–85 °C; <sup>1</sup>H NMR  $\delta$  1.13 (s, 9H), 1.58 (tm, J = 11.5, 2H), 1.85–2.10 (m, 6H), 4.49 (t, J = 4.4, 1H), 7.26 (overlap with solvent, 1H), 7.71 (d, *J* = 8.0, 1H), 8.48 (br, 1H), 8.64 (br, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>) δ 27.7 (CH<sub>3</sub>), 29.3 (CH<sub>2</sub>), 37.4 (CH<sub>2</sub>), 59.8 (CH), 68.4 (C), 79.7 (C), 122.8 (CH), 134.5 (CH), 138.3 (C), 147.3 (CH), 148.0 (CH), 156.8 (C). Anal. Calcd for C<sub>16</sub>H<sub>22</sub>N<sub>2</sub>O<sub>2</sub>: C, 70.0; H, 8.1; N, 10.2. Found: C, 69.8; H, 8.2; N, 9.9.

1-(3-Pyridinyl)-7-azabicyclo[2.2.1]heptane (2b). To 2a (39 mg, 0.14 mmol) was added a solution of HCl in EtOAc (3 M, 2 mL), the resulting solution was stirred at room temperature for 1 h and then evaporated. The residue was distributed between saturated aqueous  $Na_2CO_3$  (10 mL) and EtOAc (2  $\times$ 10 mL), the combined organic layer was dried and filtered, and the filtrate was evaporated to afford 2b (24 mg, 95%) as a colorless oil: <sup>1</sup>H NMR  $\delta$  1.70–2.10 (m, 8H), 3.15 (br, 1H), 3.82 (t, J = 4.7, 1H), 7.26 (dd, J = 7.9, 4.8, 1H), 7.79 (dt, J =7.9, 1.7, 1H), 8.50 (dd, J = 4.8, 1.6, 1H), 8.71 (d, J = 1.6, 1H);  $^{13}\mathrm{C}$  NMR (CDCl\_3)  $\delta$  31.8 (CH2), 37.3 (CH2), 57.0 (CH), 67.5 (C),123.0 (CH), 133.4 (CH), 139.3 (C), 147.5 (CH), 147.9 (CH). The hydrogen fumarate of 2b was formed by mixing 2b (20 mg, 0.115 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (2 mL) with fumaric acid (15 mg, 0.129 mmol) in MeOH (2.47 mL) and evaporating the solution to an amorphous white solid (35 mg, 99%): <sup>1</sup>H NMR (CD<sub>3</sub>OD)  $\delta$  1.99 (tm, J = 11.5, 2H), 2.05–2.30 (m, 4H), 2.38 (tm, J =11.5, 2H), 4.38 (t, J = 4.7, 1H), 6.67 (s, 2H), 7.54 (dd, J = 8.0, 4.7, 1H), 7.98 (d, J = 8.0, 1H), 8.59 (br, 1H), 8.68 (bs, 1H); <sup>13</sup>C NMR (CD<sub>3</sub>OD) δ 29.3 (CH2), 35.3 (CH2), 60.8 (CH), 73.0 (C),-125.5 (CH), 134.8 (C), 135.8 (CH), 136.1 (CH), 147.4 (CH),-150.4 (CH), 171.0 (C). Anal. Calcd. for C<sub>11</sub>H<sub>14</sub>N<sub>2</sub>·1.1C<sub>4</sub>H<sub>4</sub>O<sub>4</sub>: C, 61.3; H, 6.1; N, 9.3. Found: C, 61.1; H, 6.4; N, 9.2.

**7-Methyl-1-(3-pyridinyl)-7-azabicyclo[2.2.1]heptane (2).** A mixture of **2b** (50 mg) in H<sub>2</sub>O (0.5 mL), formic acid (0.25 mL), and formaldehyde (0.25 mL) was heated at reflux for 14 h. The solution was evaporated, the residue was partitioned between EtOAc (10 mL) and 2 M  $K_2CO_3$  (10 mL), and the aqueous layer was extracted with EtOAc (2  $\times$  10 mL). The combined organic layer was dried, filtered, and evaporated, and the residue was chromatographed (CH<sub>2</sub>Cl<sub>2</sub>/MeOH saturated with NH<sub>3</sub>, 97/3) to afford **2** (51 mg, 94%) as a colorless oil: <sup>1</sup>H NMR  $\delta$  1.30–2.80 (br, 11H, with s at 2.04, 3H), 3.46 (t, J= 4.3, 1H), 7.28 (dd, J= 7.9, 4.8, 1H), 7.78 (dt, J= 7.9, 1.7, 1H), 8.50 (d, J= 4.8, 1H), 8.62 (s,1H); <sup>13</sup>C NMR (two CH<sub>2</sub> obscured by pyramidyl inversion)  $\delta$  32.3, 63.2, 70.1, 123.4, 135.2, 137.7, 148.2, 149.0. Anal. Calcd for C<sub>11</sub>H<sub>14</sub>N<sub>2</sub>·0.5H<sub>2</sub>O: C, 73.1; H, 8.7; N, 14.2. Found: C, 73.2; H, 8.7; N, 13.9.

(2S,5R)-1-(tert-Butyloxycarbonyl)-2-hydroxymethyl-5-(3-pyridinyl)pyrrolidine ((+)-25). To a solution of (+)-15 (400 mg, 1.3 mmol) in EtOH (40 mL) at 0 °C was added CaCl<sub>2</sub> (433 mg, 3.9 mmol) followed by NaBH<sub>4</sub> (296 mg, 7.8 mmol). The mixture was stirred overnight as it warmed to room temperature. Aqueous K<sub>2</sub>CO<sub>3</sub> (2 M, 5 mL) was added, the mixture was partitioned between EtOAc (50 mL) and saturated aqueous NaHCO3 (20 mL), the aqueous layer was extracted with EtOAc ( $2 \times 40$  mL), and the combined organic layer was dried, filtered, and evaporated. The residue was chromatographed (EtOAc) to afford 25 (315 mg, 87%) as a white solid: mp 75–78 °C;  $[\alpha]^{22}_{D}$  +17.4 (*c* 1, CHCl<sub>3</sub>); <sup>1</sup>H NMR δ 1.20 (bs, 9H), 1.69 (br, 1H), 1.87 (m, 1H), 2.07 (m, 1H), 2.34 (m, 1H), 3.80 (m, 2H), 4.18 (br, 1H), 4.55 (br, 1H), 4.83 (bt, 1H), 7.6 (m, 1H), 7.39 (d, J = 4.8, 1H), 8.52 (m, 2H); <sup>1</sup>H NMR  $(C_6D_6, 298 \text{ K}, \text{ rotamers}) \delta 1.07 \text{ (bs, 9H)}, 1.20-1.70 \text{ (bm, 4H)},$ 3.60-3.85 (br, 2H), 4.21 (br, 2H), 4.66 (br, 1H), 6.75 (dd, J= 7.9, 4.7, 1H), 7.13 (br, 1H), 8.42 (bd, J = 4.7, 1H), 8.51 (br, 1H); <sup>1</sup>H NMR (C<sub>6</sub>D<sub>6</sub>, 329 K) δ 1.22 (s, 9H), 1.25–1.38 (m, 1H), 1.40-1.72 (m, 4H), 1.75-1.83 (m, 1H), 3.70-3.90 (m, 2H), 3.80-4.20 (br, 1H), 4.28 (bq, J = 7.0, 1H), 4.45 (dt, J = 7.4, 1H), 6.89 (bd, J = 7.9, 4.7, 1H), 7.31 (d, J = 7.9, 1H), 8.51 (d, J = 4.7, 1H), 8.63 (s, 1H); <sup>13</sup>C NMR (C<sub>6</sub>D<sub>6</sub>, 328 K)  $\delta$  27.2, 28.1, 34.1, 61.2, 61.6, 65.9, 80.0, 123.1, 132.9, 140.1, 148.4, 148.7, 156.0. Anal. Calcd for  $C_{15}H_{22}N_2O_3$ : C, 64.7; H, 8.0; N, 10.1. Found: C, 65.0; H, 8.0; N, 10.4.

(2*R*,5*.***S**)-1-(*tert*-Butyloxycarbonyl)-2-hydroxymethyl-5-(3-pyridinyl)pyrrolidine ((–)-25, structure not shown) was prepared from (–)-15 in the same way:  $[\alpha]^{22}_D - 17.0$  (*c* 1, CHCl<sub>3</sub>); spectral and chromatographic properties identical with those of (+)-25.

(2S,5R)-1-(tert-Butyloxycarbonyl)-2-formyl-5-(3-pyridinyl)pyrrolidine (26) and (2S,5R)-1-(tert-Butyloxycarbonyl)-2-(2-methoxycarbonylvinyl)-5-(3-pyridinyl)pyrroli**dine (27).** To a solution of oxalyl chloride (0.75 g, 5.9 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (10 mL) at -78 °C was added DMSO (0.84 mL, 11.8 mmol), the mixture was stirred for 15 min, a solution of 25 (0.82 g, 2.95 mmol) in  $CH_2Cl_2$  (10 mL) was added over 10 min, the mixture was stirred for 1 h, and triethylamine (2.5 mL, 17.7 mmol) was added dropwise over 5 min. The mixture was stirred for an additional 1 h and poured into  $H_2O$  (30 mL). The organic layer was washed with H<sub>2</sub>O (20 mL) and brine (20 mL), dried, filtered, and evaporated to afford crude aldehyde **26**: <sup>1</sup>H NMR (298 K, rotamers)  $\delta$  1.00–1.50 (br, 9H), 1.70-2.45 (bm, 4H), 4.26 (br, 0.5H), 4.46 (br, 0.5H), 4.78 (br, 0.5H), 5.02 (br, 0.5H), 7.26 (br, 1H), 7.60-7.80 (br, 1H), 8.40-8.70 (br, 2H). To a suspension of NaH (85 mg, 3.54 mmol) in THF (40 mL) at 0  $^{\circ}C$  was added a solution of trimethyl phosphonoacetate (0.57 mL, 3.54 mmol) dropwise, the resulting mixture was stirred at 0 °C for 30 min, and the crude 26 in THF (40 mL) was added via cannula. The resulting solution was stirred for another 30 min, pH 7 phosphate buffer (5 mL) was added, the mixture was concentrated to about 20 mL, and the residue was partitioned between EtOAc (25 mL) and H<sub>2</sub>O (25 mL). The aqueous phase was extracted with EtOAc (2 imes25 mL), and the combined organic phase was dried and evaporated, leaving a residue that was chromatographed (EtOAc/hexanes, 6/4) to give the unsaturated ester 27 (0.73 g, 73% from **25**) as a mixture of cis and trans isomers: <sup>1</sup>H NMR (298 K)  $\delta$  1.10–1.51 (br, 9H major, 9H minor), 1.68 (br, 1H minor), 1.86 (m, 2H major, 1H minor), 2.13 (m, 1H major), 2.30 (m, 1H major, 2H minor), 3.67 (s, 3H minor), 3.71 (s, 3H major), 4.30-4.70 (br, 1H major), 4.70-5.10 (br, 1H major, 1H minor), 5.43 (bq, J = 7.2, 1H minor), 5.80 (bd, J = 11.1, 1H minor), 6.00 (br, 1H major), 6.23-6.57 (br, 1H minor), 6.95 (br, 1H major), 7.22 (m, 1H major, 1H minor), 7.53 (m, 1H major, 1H minor), 8.45 (m, 1H major, 1H minor), 8.49 (s, 1H major), 8.51 (s, 1H minor); <sup>1</sup>H NMR (328 K) & 1.27 (s, 9H major), 1.28 (s, 9H minor), 1.68 (m, 1H minor), 1.90 (m, 2H major, 1H minor), 2.15 (m, 1H major), 2.31 (m, 1H major, 2H minor), 3.69 (s, 3H minor), 3.72 (s, 3H major), 4.55 (br, 1H major), 4.85 (br, 1H major), 4.96 (br, 1H minor), 5.44 (bq, J = 7.2, 1H minor), 5.80 (dd, J = 11.5, 1.2, 1H minor), 6.00 (d, J = 15.7, 1H major), 6.37 (bt, J = 9.6, 1H minor), 6.96 (dd, J = 15.7, 6.5, 1H major), 7.21 (m, 1H major, 1H minor), 7.53 (m, 1H major, 1H minor), 8.44 (d, J = 4.7, 1H major, 1H minor), 8.51 (s, 1H major), 8.53 (s, 1H minor); <sup>13</sup>C NMR (328 K)  $\delta$  28.1  $\times$  2, 30.4, 30.5, 33.6, 33.9, 51.0, 51.4, 57.4, 59.5, 60.4, 60.8, 80.2, 80.4, 118.5, 121.5,123.1, 123.2, 133.2, 138.8, 139.1, 147.8, 148.1  $\times$  2, 148.2  $\times$  2, 152.2, 154.6, 154.8, 166.1, 166.5. Anal. Calcd for C18H24N2O4: C, 65.0; H, 7.3; N, 8.4. Found: C, 64.7; H, 7.5; N, 8.2.

(2S,5R)-1-(tert-Butyloxycarbonyl)-2-(2- $\beta$ -methoxycarbonylethyl)-5-(3-pyridinyl)pyrrolidine (28). Unsaturated ester 27 (1.0 g, 3.0 mmol) in MeOH (10 mL) was treated with  $H_2$  at 1 atm in the presence of 10% Pd/C (100 mg, 10% w/w) at room temperature for 4 h. The resulting mixture was filtered through Celite and evaporated to give 28 (1.0 g, 99%) as a colorless oil: <sup>1</sup>H NMR (298 K, rotamers)  $\delta$  1.00–1.50 (br, 9H), 1.65 (m, 1H), 1.70-1.92 (m, 2H), 2.00 (m, 1H), 2.13-2.37 (m, 2H), 2.41 (br, 1H), 3.64 (s, 3H), 3.97 (br, 1H), 4.60-5.20 (br, 1H), 7.21 (dd, J = 7.8, 4.7, 1H), 7.51 (d, J = 7.8, 1H), 8.43 (d, J = 4.8, 1H), 8.45 (s, 1H); <sup>1</sup>H NMR (328 K)  $\delta$  1.26 (s, 9H), 1.68 (m, 1H), 1.80 (m, 1H), 1.88 (m, 1H), 2.07 (m, 1H), 2.15-2.35 (m, 2H), 2.41 (t, J = 7.8, 2H), 3.65 (s, 3H), 4.00 (m, 1H), 4.77 (t, J = 7.6, 1H), 7.19 (dd, J = 7.8, 4.8, 1H), 7.52 (dt, J = 7.8,1.8, 1H), 8.43 (d, J = 4.8, 1H), 8.48 (d, J = 1.8, 1H); <sup>13</sup>C NMR (328 K) & 28.2, 29.8, 31.0, 31.5, 33.9, 51.3, 58.6, 60.8, 79.9, 123.1, 133.1, 139.6, 147.9, 148.0, 154.9, 173.4.

(2S,5R)-1-(tert-Butyloxycarbonyl)-2-(3-hydroxypropyl)-5-(3-pyridinyl)pyrrolidine ((+)-29). To a solution of ester 28 (1 g, 3 mmol) in EtOH (30 mL) at 0 °C was added CaCl<sub>2</sub> (1 g, 9 mmol) followed by NaBH<sub>4</sub> (0.68 g, 18 mmol). The mixture was stirred overnight as it warmed to room temperature, aqueous K<sub>2</sub>CO<sub>3</sub> (2 M, 20 mL) was added, and the mixture was evaporated. The residue was partitioned between EtOAc (100 mL) and H<sub>2</sub>O (50 mL), the aqueous layer was extracted with EtOAc (2  $\times$  100 mL), and the combined organic layer was dried, filtered, and evaporated. The residue was chromatographed (EtOAc) to afford alcohol 29 (0.65 g, 71%) as a colorless oil:  $[\alpha]^{22}_{D}$  +16.5 (c 1, CHCl<sub>3</sub>); <sup>1</sup>H NMR (298 K)  $\delta$ 1.18-1.45 (br s, 9H), 1.45-1.76 (m, 4H), 1.87 (m, 1H), 2.00 (m, 2H), 2.26 (m, 1H), 2.40-3.50 (br, 1H), 3.67 (bs, 2H), 3.98 (br, 1H), 4.69 (br, 1H), 7.19 (dd, J = 7.8, 4.8, 1H), 7.51 (dt, J = 7.8, 1.8, 1H), 8.41 (d, J = 4.8, 1H), 8.45 (d, J = 1.8, 1H); <sup>1</sup>H NMR (328 K)  $\delta$  1.23 (s, 9H), 1.48–1.72 (m, 4H), 1.87 (m, 1H), 2.01 (m, 2H), 2.28 (m, 1H), 2.66 (br, 1H), 3.69 (bs, 2H), 4.02 (br, 1H), 4.74 (dt, J = 7.2, 1H), 7.18 (dd, J = 7.9, 4.7, 1H), 7.51 (d, J = 7.9, 1.8, 1H), 8.42 (dd, J = 4.7, 1.8, 1H), 8.47 (d, J = 1.8, 1H); <sup>13</sup>C NMR (328 K)  $\delta$  28.2, 29.7, 29.8, 32.1, 34.1, 58.9, 60.7, 62.4, 79.7, 123.1, 133.1, 140.0, 147.9, 154.9. Anal. Calcd for C<sub>17</sub>H<sub>26</sub>N<sub>2</sub>O<sub>3</sub>: C, 66.6; H, 8.6; N, 9.1. Found: C, 66.4; H, 8.8; N, 9.0.

(2*R*,5*.***S**)-1-(*tert*-Butyloxycarbonyl)-2-(3-hydroxypropyl)-5-(3-pyridinyl)pyrrolidine ((-)-29, structure not shown) was prepared in the same manner from ester (-)-28:  $[\alpha]^{22}_{\rm D}$ -17.0 (*c* 1.2, CHCl<sub>3</sub>); spectral and chromatographic properties identical with (+)-29.

(2.5,5*R*)-1-(*tert*-Butyloxycarbonyl)-2-(3-methylsulfonyloxypropyl)-5-(3-pyridinyl)pyrrolidine (30) and (2.5,5*R*)-1-(*tert*-Butyloxycarbonyl)-2-(3-bromopropyl)-5-(3-pyridinyl)pyrrolidine (31). To a solution of 1-methansulfonylimidazolide (180 mg, 1.2 mmol) in THF (12 mL) at 0 °C was added methyl triflate (127  $\mu$ L, 1.2 mmol) dropwise, and the mixture was stirred at 0 °C for 30 min followed by the dropwise addition at 0 °C of a mixture of **29** (315 mg, 1.03 mmol) and 1-methylimidazole (84  $\mu$ L, 1.03 mmol) in THF (3 mL). After the addition, the ice bath was removed, the mixture was stirred

overnight and then evaporated, and the residue was partitioned between EtOAc (2  $\times$  20 mL) and H<sub>2</sub>O (20 mL). The organic layer was dried, filtered, and evaporated to afford crude mesylate 30 containing a small amount of mesylimidazolide: <sup>1</sup>H NMR (298 K, rotamers)  $\delta$  1.00–2.40 (m, 17H), 2.98 (s, 3H), 3.95 (bs, 1H), 4.28 (bs, 2H), 4.75 (br, 1H), 7.20 (br, 1H), 7.49 (bd, J = 7.9, 1H), 8.48 (bs, 2H). Crude **30** was dissolved in acetone (12 mL), LiBr (358 mg, 4.12 mmol) was added, and the mixture was refluxed for 1 h. Evaporation left a residue that was partitioned between EtOAc (10 mL) and  $H_2O$  (5 mL). The aqueous layer was extracted with EtOAc (2  $\times$  10 mL), and the combined organic layer was dried, filtered, and evaporated. The residue was chromatographed (CH<sub>2</sub>Cl<sub>2</sub>/ EtOAc, 4/6) to afford bromide 31 (270 mg, 73%) as a colorless oil: <sup>1</sup>H NMR (298 K, rotamers) δ 1.00–1.50 (br, 9H), 1.55– 1.78 (br, 2H), 1.85-2.10 (m, 5H), 2.20-2.28 (m, 1H), 3.44 (m, 2H), 4.02 (br, 1H), 4.60–5.00 (br, 1H), 7.27 (dd, J = 7.9, 4.7, 1H), 7.57 (d, J = 7.9, 1H), 8.47 (d, J = 4.7, 1H), 8.57 (s, 1H); <sup>1</sup>H NMR (323 K) δ 1.28 (bs, 9H), 1.55-1.78 (m, 2H), 1.85-2.10 (m, 5H), 2.20-2.28 (m, 1H), 3.44 (m, 2H), 4.02 (br, 1H), 4.79 (bt, 1H), 7.23 (dd, J = 7.9, 4.7, 1H), 7.54 (d, J = 7.9, 1H), 8.47 (d, J = 4.7, 1H), 8.54 (s, 1H); <sup>13</sup>C NMR (328 K)  $\delta$  28.2, 30.1, 30.3, 33.3, 34.0, 34.5, 58.4, 61.8, 79.9, 123.1, 133.2, 139.6, 147.8, 147.9, 154.8. Bromide 31 is unstable and was used immediately after a short period of drying under vacuum.

8-tert-Butyloxycarbonyl-1-(3-pyridinyl)-8-azabicyclo-[3.2.1]octane ((-)-3a). To a solution of *n*-BuLi (0.21 mmol, 1.6 M in hexane, 0.13 mL) in THF (15 mL) at -78 °C was added bromide **31** (27 mg, 0.083 mmol) in THF (2 mL) slowly (syringe pump) over 1 h. The mixture was stirred at -78 °C for 30 min, and then pH 7 phosphate buffer (20 mL) was added and the reaction mixture was allowed to warm to room temperature. The mixture was extracted with EtOAc (2  $\times$  20 mL), and the combined organic layer was dried, filtered, and evaporated. The residue was chromatographed (hexanes/ EtOAc, 1/1) to afford 3a (10 mg, 48%) as a white solid: mp 104–106 °C;  $[\alpha]^{22}_{D}$  –46.1 (*c* 1, ČHCl<sub>3</sub>); <sup>1</sup>H NMR  $\delta$  1.50–2.40 (m, 10H), 3.49 (s, 3H), 4,47 (bs, 1H), 7.22 (dd, J = 7.8, 4.8, 1H), 7.63 (d, J = 7.8, 1H), 8.43 (bs, 1H), 8.60 (s, 1H); <sup>13</sup>C NMR δ 17.6, 26.7, 27.9, 29.0, 30.9, 41.9, 57.8, 63.7, 79.4, 122.7, 132.9, 141.9, 147.0, 154.4. Anal. Calcd for C17H24N2O2: C, 70.8; H, 8.4; N, 9.7. Found: C, 70.4; H, 8.8; N, 9.7.

**8**-(*tert*-Butoxycarbonyl)-1-(3-pyridinyl)-8-azabicyclo-[3.2.1]octane ((+)-3a) was prepared from the enantiomeric substrate in the same manner:  $[\alpha]^{22}_D$  +45.8 (*c* 0.93, CHCl<sub>3</sub>); spectral and chromatographic properties identical with those of (-)-3a.

**1-(3-Pyridinyl)-8-azabicyclo[3.2.1]octane ((–)-3b)**. To **3a** (39 mg, 0.14 mmol) was added HCl in EtOAc (3 M, 2 mL), and the resulting mixture was stirred at room temperature for 30 min. After evaporation, the residue was partitioned between saturated aqueous Na<sub>2</sub>CO<sub>3</sub> (10 mL) and EtOAc (2 × 10 mL). The combined organic layer was dried, filtered, and evaporated to afford (–)-**3b** (23 mg, 93%) as a colorless oil:  $[\alpha]^{22}_{D}$  –7.0 (*c* 1.1, CHCl<sub>3</sub>); <sup>1</sup>H NMR  $\delta$  1.50–2.10 (m, 10H), 2.20–2.32 (m, 1H), 3.72 (m, 1H), 7.24 (dd, J = 7.9, 4.8, 1H), 7.73 (dt, J = 7.9, 1.7, 1H), 8.46 (dd, J = 4.8, 1.6, 1H), 8.65 (d, J = 1.6, 1H); <sup>13</sup>C NMR  $\delta$  18.7, 30.1, 31.6, 35.8, 41.1, 55.5, 64.1, 123.1, 132.8, 143.7, 146.3, 147.6. Anal. Calcd for Cl<sub>2</sub>H<sub>16</sub>N<sub>2</sub>-<sup>1</sup>/<sub>4</sub>H<sub>2</sub>O: C, 74.8; H, 8.6; N, 14.5. Found: C, 75.1; H, 8.8; N, 14.7.

**1-(3-Pyridinyl)-8-azabicyclo[3.2.1]octane ((+)-3b)** was prepared from the enantiomeric substrate in the same manner:  $[\alpha]^{22}_{D}$  +6.8 (c 0.96, CHCl<sub>3</sub>); spectral and chromatographic properties identical with those of (-)-**3b**.

7-Methyl-1-(3-pyridinyl)-8-azabicyclo[3.2.1]octane ((–)-3). A mixture of (–)-3b (50 mg) in H<sub>2</sub>O (0.5 mL), formic acid (0.25 mL), and formaldehyde (0.25 mL) was heated at reflux for 14 h and then evaporated, and the residue was partitioned between EtOAc (5 mL) and 2 M K<sub>2</sub>CO<sub>3</sub> (5 mL). The aqueous layer was extracted with EtOAc (2 × 5 mL), and the combined organic layer was dried and evaporated. The residue was chromatographed (CH<sub>2</sub>Cl<sub>2</sub>/MeOH saturated with NH<sub>3</sub>, 97/3) to afford (–)-3 (48 mg, 90%) as a colorless oil:  $[\alpha]^{22}_{\rm D}$ –10.9

7-Methyl-1-(3-pyridinyl)-8-azabicyclo[3.2.1]octane ((+)-3) was prepared from the enantiomeric substrate in the same

manner:  $[\alpha]^{22}_{D}$  +11.2 (*c* 1.3, CHCl<sub>3</sub>); spectral and chromatographic properties identical with those of (-)-**3**.

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